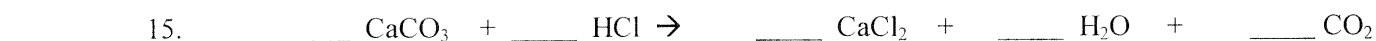
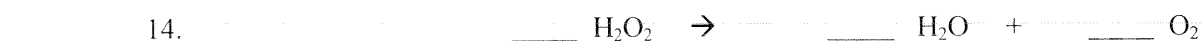
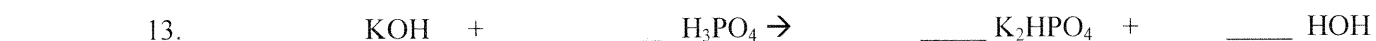
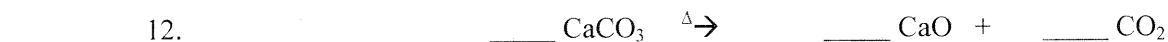
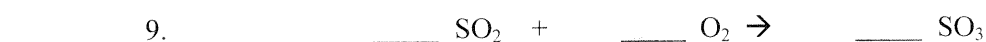
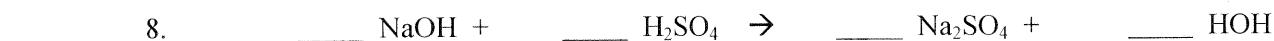
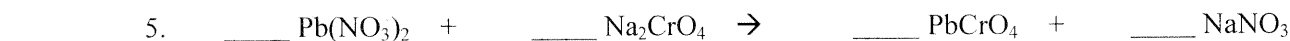
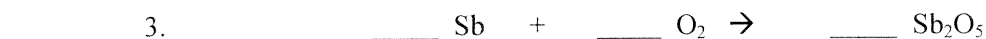


Name: _____

Hour: _____

Balancing Equations: Extra Practice

Directions: Balance the following equations and classify each as composition, decomposition, single replacement or double replacement.



Directions: Write AND BALANCE the following equations:

16) **Mg** reacts with **HCl** to produce **MgCl₂** and **H₂**

17) **Ag(NO₃)** reacts with **Ca** to produce **Ca(NO₃)₂** and **Ag**

18) **H(NO₃)** reacts with **Na(OH)** to produce **NaNO₃** and **H(OH)**

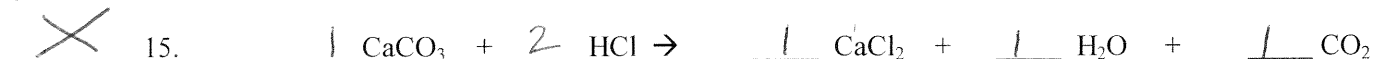
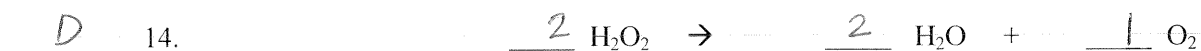
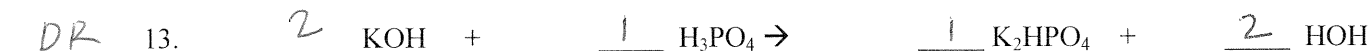
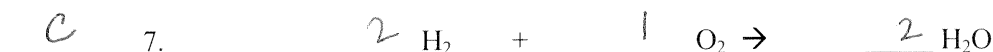
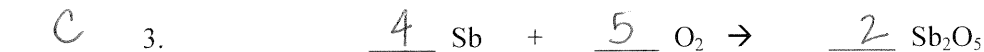
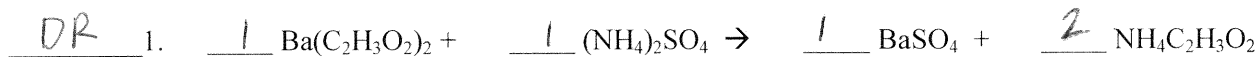
Name: _____

Hour: _____

Balancing Equations: Extra Practice

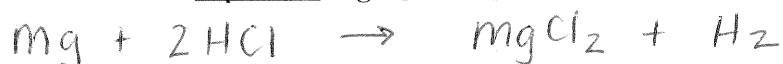
KEY

Directions: Balance the following equations and classify each as composition, decomposition, single replacement or double replacement.

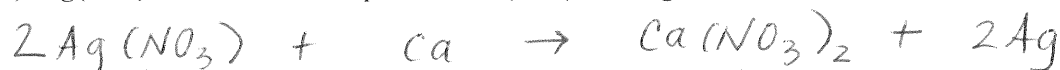


Directions: Write AND BALANCE the following equations:

16) Mg reacts with HCl to produce MgCl₂ and H₂



17) Ag(NO₃) reacts with Ca to produce Ca(NO₃)₂ and Ag



18) H(NO₃) reacts with Na(OH) to produce NaNO₃ and H(OH)

